

6 α -AcetoxypoxyazadiradioneWisanu Maneerat,^a Hoong-Kun Fun,^{b*} Taweesak Buatip,^a Surat Laphookhieo^a and Suchada Chantrapromma^{c*}^aSchool of Science, Mae Fah Luang University, Tasud, Muang Chiang Rai 57100, Thailand, ^bX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, and ^cDepartment of Chemistry, Faculty of Science, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand

Correspondence e-mail: hkfun@usm.my, suchada.c@psu.ac.th

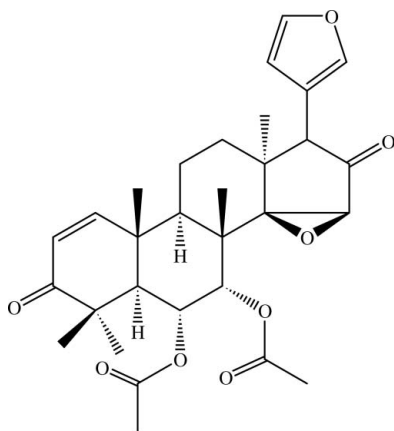
Received 15 July 2007; accepted 18 July 2007

Key indicators: single-crystal X-ray study; $T = 297$ K; mean $\sigma(\text{C}-\text{C}) = 0.006$ Å; R factor = 0.051; wR factor = 0.125; data-to-parameter ratio = 7.8.

The title limonoid compound [systematic name: (8*S**,10*R**)-17-(3-furyl)-4,4,8,10,13-pentamethyl-3,16-dioxo-6,7,9,11,12,17-hexahydro-14,15-epoxy-5*H*-cyclopenta[*a*]phenanthrene-6 α ,7 α -diyl diacetate], C₃₀H₃₆O₈, was isolated from *Chisocheton siamensis*. The molecule contains four *trans*-fused rings. The cyclohexene ring adopts a screw-boat conformation, the two cyclohexane rings are in standard chair and boat conformations, and the cyclopentane ring is in an envelope conformation. Weak intramolecular C—H \cdots O interactions generate *S*(5) and *S*(6) ring motifs. A weak C—H \cdots O intermolecular interaction connects the molecules into dimers.

Related literature

For hydrogen-bond motifs, see Bernstein *et al.* (1995). For values of bond lengths, see Allen *et al.* (1987). For ring conformations, see Cremer & Pople (1975). For limonoid compounds and activities, see, for example, Bickii *et al.* (2000); Khalid *et al.* (1998); Koul *et al.* (2003); Nihei *et al.* (2002); Penido, Conte *et al.* (2006); Penido, Costa *et al.* (2006); Saewan *et al.* (2006); Takeya *et al.* (1996).



Experimental

Crystal data

C₃₀H₃₆O₈
 $M_r = 524.59$
 Orthorhombic, $P2_12_12$
 $a = 12.189$ (3) Å
 $b = 29.983$ (5) Å
 $c = 7.3372$ (11) Å
 $V = 2681.5$ (9) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 297$ (2) K
 $0.39 \times 0.10 \times 0.07$ mm

Data collection

Bruker SMART APEXII CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2005)
 $T_{\min} = 0.965$, $T_{\max} = 0.994$
 20575 measured reflections
 2724 independent reflections
 1782 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.100$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.125$
 $S = 1.02$
 2724 reflections
 351 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-----------------------------------|-------|-------------|-------------|---------------|
| C5—H5A \cdots O7 | 0.98 | 2.45 | 2.900 (5) | 107 |
| C7—H7A \cdots O8 | 0.98 | 2.31 | 2.705 (5) | 103 |
| C9—H9A \cdots O7 | 0.98 | 2.54 | 2.936 (5) | 104 |
| C27—H27A \cdots O2 ⁱ | 0.96 | 2.51 | 3.422 (7) | 160 |
| C28—H28D \cdots O5 | 0.96 | 2.53 | 3.121 (6) | 120 |
| C29—H29D \cdots O5 | 0.96 | 2.29 | 2.973 (5) | 127 |

Symmetry code: (i) $-x, -y, z$.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

SL, WM and TB thank Mae Fah Luang University and the Bioresources Research Network, National Center for Genetic and Engineering and Biotechnology (grant No. BRN 003 G-49), for partial financial support. The authors also thank the Malaysian Government and Universiti Sains Malaysia for Scientific Advancement Grant Allocation (SAGA) grant No. 304/PFIZIK/635003/A118.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2461).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bickii, J., Njifutie, N., Foyere, J. A., Basco, L. K. & Ringwald, P. (2000). *J. Ethnopharmacol.* **69**, 27–33.
 Bruker (2005). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.

- Khalid, S. A., Friedrichsen, G. M., Kharazmi, A., Theander, T. G., Olsen, C. E. & Christensen, S. B. (1998). *Phytochemistry*, **49**, 1769–1772.
- Koul, O., Daniewski, W. M., Multani, J. S., Gumulka, M. & Singh, G. (2003). *J. Agric. Food Chem.* **51**, 7271–7275.
- Nihei, K.-I., Hanke, F. J., Asaka, Y., Matsumoto, T. & Kubo, I. (2002). *J. Agric. Food Chem.* **50**, 5048–5052.
- Penido, C., Conte, F., Chagas, M., Rodrigues, C., Pereira, J. & Henriques, M. (2006). *Int. Immunopharmacol.* **6**, 109–121.
- Penido, C., Costa, K. A., Costa, M., Pereira, J. & Henriques, M. (2006). *Inflamm. Res.* **55**, 457–464.
- Saewan, N., Sutherland, J. D. & Chantrapromma, K. (2006). *Phytochemistry*, **67**, 2288–2293.
- Sheldrick, G. M. (1998). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Takeya, K., Qiao, Z. S., Hirobe, C. & Itokawa, H. (1996). *Phytochemistry*, **42**, 709–712.

supplementary materials

Acta Cryst. (2007). E63, o3583–o3584 [doi:10.1107/S1600536807035155]

6*O*-Acetoxypoxyazadiradione

W. Maneerat, H.-K. Fun, T. Buatip, S. Laphookhieo and S. Chantrapromma

Comment

Meliaceae plants are known to be rich sources of limonoids. Some of these limonoids possess significant pharmacological effects such as antimalarial (Bickii *et al.*, 2000; Saewan *et al.*, 2006), cytotoxic (Takeya *et al.*, 1996), antiprotozoal (Khalid *et al.*, 1998) and antifeedant activities (Koul *et al.*, 2003; Nihei *et al.*, 2002). *Chisocheton siamensis* or "Ta Sua" in a local Thai name, is one of the Meliaceae plants, which is found in the northern part of Thailand. As part of our study of chemical constituents and bioactive compounds from *C. siamensis*, we herein report the crystal structure of the title compound which was isolated for the first time from the seed of *C. siamensis* collected from Chiang Mai province in the northern part of Thailand. The title compound was found to possess anti-allergic (Penido, Costa *et al.*, (2006) and antiinflammatory (Penido, Conte *et al.*, (2006) activities.

The title molecule (Fig. 1) contains a four *trans*-fused rings, the cyclohexene ring adopts a screw boat conformation with the puckering parameters (Cremer & Pople, 1975) $Q = 0.552(4) \text{ \AA}$, $\theta = 70.6(5)^\circ$ and $\varphi = 276.7(5)^\circ$, the two cyclohexane rings are in standard chair and boat conformations and the cyclopentane adopts an envelope conformation, with atom C17 displaced from the C13/C14/C15/C16 plane by $0.238(4) \text{ \AA}$. The oxirane moiety (C14/C15/O3) makes the dihedral angle of $78.0(4)^\circ$ with the mean plane of C13/C14/C15/C16. The furan ring is planar and attached to the cyclopentane ring at atom C17, the dihedral angle between the furan ring and the C13/C14/C15/C16 mean plane is $56.7(3)^\circ$. The two acetate groups are planar and in *cis*-configuration with respect to the C6—C7 bond, the dihedral angle between the mean plane of these two acetate moieties is $44.4(3)^\circ$ (Fig. 1). The bond lengths and angles are within normal ranges (Allen *et al.*, 1987). In the structure, the weak C5—H5A...O7, C7—H7A...O8 and C9—H9A...O7 intramolecular interactions generate S(5) ring motifs, the C28—H28D...O5 and C29—H28D...O5 generate S(6) ring motifs (Bernstein *et al.*, 1995), whereas the intermolecular C27—H27A...O2($-x, -y, z$) interaction connects the molecules into dimers (Fig. 2).

Experimental

Seeds of *C. siamensis* (600 g) were extracted with hexane-acetone (1:1 V/V) over the period of 3 days at room temperature. The mixture was filtered and concentrated under reduced pressure to provide the crude extract (62.21 g). This crude extract was subjected to quick column chromatography (QCC) over silica gel and eluted with a gradient of EtOAc-hexane to afford 13 fractions (A1—A13). Fraction A8 (1.01 g) was subjected to repeated column chromatography using 25% EtOAc-hexane to give compound (I) (17.2 mg). Colorless plate shaped single crystals of the title compound were recrystallized from EtOAc-hexane (1:2 V/V) solution, Mp 493 K decomposition.

Refinement

All H atoms were positioned geometrically and allowed to ride on their parent atoms, with the C—H distances in the range 0.93–0.98 Å. The U_{iso} values were constrained to be $1.5U_{eq}$ of the carrier atom for methyl H atoms and $1.2U_{eq}$ for the remaining H atoms. A rotating group model was used for the methyl groups. A total of 2006 Friedel pairs were merged

supplementary materials

before final refinement as there is no large anomalous dispersion for the determination of the absolute configuration. The absolute configuration of the title compound was selected arbitrarily.

Figures

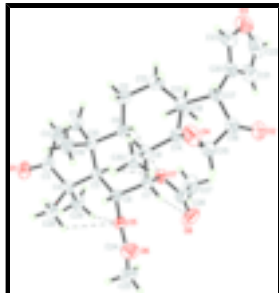


Fig. 1. The asymmetric unit of showing 40% probability displacement ellipsoids and the atomic numbering scheme. Intramolecular hydrogen bonds are shown as dashed lines.

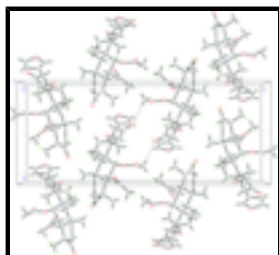


Fig. 2. The crystal packing viewed along the *c* axis. Weak C—H...O interactions are shown as dash lines.

(8*S,10*R**)-17-(3-furyl)-4,4,8,10,13-pentamethyl-3,16-dioxo-6,7,9,11,12,17-hexahydro-14,15-epoxy-5*H*-cyclopenta[*a*]phenanthrene-6*α*,7*α*-diyl diacetate**

Crystal data

$C_{30}H_{36}O_8$

$M_r = 524.59$

Orthorhombic, $P2_12_12$

Hall symbol: P 2 2ab

$a = 12.189$ (3) Å

$b = 29.983$ (5) Å

$c = 7.3372$ (11) Å

$V = 2681.5$ (9) Å³

$Z = 4$

$F_{000} = 1120$

$D_x = 1.299$ Mg m⁻³

Melting point: 493 K, decomposition K

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 2724 reflections

$\theta = 1.8$ – 25.0°

$\mu = 0.09$ mm⁻¹

$T = 297$ (2) K

Plate, colourless

$0.39 \times 0.10 \times 0.07$ mm

Data collection

Bruker SMART APEXII CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 8.33 pixels mm⁻¹

$T = 297$ (2) K

2724 independent reflections

1782 reflections with $I > 2\sigma(I)$

$R_{int} = 0.100$

$\theta_{max} = 25.0^\circ$

$\theta_{min} = 1.8^\circ$

ω scans $h = -14 \rightarrow 12$
 Absorption correction: multi-scan $k = -35 \rightarrow 32$
 (SADABS; Bruker, 2005)
 $T_{\min} = 0.965$, $T_{\max} = 0.994$ $l = -8 \rightarrow 8$
 20575 measured reflections

Refinement

Refinement on F^2 Hydrogen site location: inferred from neighbouring sites
 Least-squares matrix: full H-atom parameters constrained
 $R[F^2 > 2\sigma(F^2)] = 0.051$ $w = 1/[\sigma^2(F_o^2) + (0.0548P)^2 + 0.359P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $wR(F^2) = 0.125$ $(\Delta/\sigma)_{\max} < 0.001$
 $S = 1.02$ $\Delta\rho_{\max} = 0.16 \text{ e } \text{\AA}^{-3}$
 2724 reflections $\Delta\rho_{\min} = -0.13 \text{ e } \text{\AA}^{-3}$
 351 parameters Extinction correction: SHELXL97,
 $F_c^* = kFc[1 + 0.001 \times Fc^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Primary atom site location: structure-invariant direct methods Extinction coefficient: 0.0128 (16)
 Secondary atom site location: difference Fourier map

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|-------------|--------------|-------------|----------------------------------|
| O1 | 0.7277 (3) | 0.20923 (12) | 0.2888 (6) | 0.0843 (11) |
| O2 | -0.0801 (3) | 0.04653 (14) | 0.1717 (6) | 0.0975 (13) |
| O3 | -0.1807 (4) | 0.1049 (2) | 0.7665 (6) | 0.1104 (15) |
| O4 | 0.0294 (3) | 0.14448 (14) | 0.0196 (5) | 0.0819 (11) |
| O5 | 0.4887 (2) | 0.10007 (9) | -0.1064 (4) | 0.0495 (7) |
| O6 | 0.4156 (3) | 0.11150 (12) | -0.3842 (5) | 0.0733 (10) |
| O7 | 0.3165 (2) | 0.08053 (8) | 0.1306 (4) | 0.0486 (7) |
| O8 | 0.2993 (4) | 0.02858 (12) | -0.0864 (6) | 0.1002 (13) |
| C1 | 0.4516 (4) | 0.20161 (14) | 0.4065 (6) | 0.0577 (12) |
| H1A | 0.4052 | 0.2069 | 0.5048 | 0.069* |
| C2 | 0.5586 (4) | 0.20273 (15) | 0.4328 (7) | 0.0631 (13) |

supplementary materials

| | | | | |
|------|-------------|--------------|-------------|-------------|
| H2A | 0.5858 | 0.2091 | 0.5484 | 0.076* |
| C3 | 0.6344 (4) | 0.19424 (15) | 0.2850 (7) | 0.0586 (13) |
| C4 | 0.5981 (3) | 0.16439 (13) | 0.1284 (6) | 0.0479 (11) |
| C5 | 0.4730 (3) | 0.15332 (13) | 0.1422 (5) | 0.0412 (9) |
| H5A | 0.4674 | 0.1289 | 0.2305 | 0.049* |
| C6 | 0.4207 (3) | 0.13562 (13) | -0.0334 (6) | 0.0425 (10) |
| H6A | 0.4157 | 0.1598 | -0.1228 | 0.051* |
| C7 | 0.3065 (3) | 0.11623 (13) | 0.0000 (6) | 0.0454 (10) |
| H7A | 0.2764 | 0.1047 | -0.1145 | 0.054* |
| C8 | 0.2303 (3) | 0.15175 (13) | 0.0774 (6) | 0.0453 (11) |
| C9 | 0.2831 (3) | 0.17297 (13) | 0.2505 (6) | 0.0488 (11) |
| H9A | 0.2926 | 0.1480 | 0.3353 | 0.059* |
| C10 | 0.4017 (3) | 0.19194 (12) | 0.2214 (6) | 0.0438 (10) |
| C11 | 0.2019 (4) | 0.20428 (15) | 0.3457 (7) | 0.0661 (13) |
| H11A | 0.2360 | 0.2158 | 0.4555 | 0.079* |
| H11B | 0.1872 | 0.2294 | 0.2661 | 0.079* |
| C12 | 0.0927 (4) | 0.18242 (15) | 0.3972 (7) | 0.0664 (13) |
| H12A | 0.0838 | 0.1841 | 0.5284 | 0.080* |
| H12B | 0.0335 | 0.1994 | 0.3423 | 0.080* |
| C13 | 0.0817 (3) | 0.13369 (15) | 0.3387 (6) | 0.0509 (11) |
| C14 | 0.1178 (3) | 0.13163 (15) | 0.1389 (6) | 0.0525 (11) |
| C15 | 0.0509 (4) | 0.0982 (2) | 0.0452 (8) | 0.0765 (16) |
| H15A | 0.0804 | 0.0794 | -0.0523 | 0.092* |
| C16 | -0.0310 (4) | 0.08116 (19) | 0.1807 (7) | 0.0706 (14) |
| C17 | -0.0398 (3) | 0.11626 (16) | 0.3268 (6) | 0.0574 (12) |
| H17A | -0.0844 | 0.1406 | 0.2773 | 0.069* |
| C18 | 0.1430 (4) | 0.10127 (17) | 0.4612 (7) | 0.0669 (14) |
| H18A | 0.2196 | 0.1087 | 0.4621 | 0.100* |
| H18B | 0.1338 | 0.0715 | 0.4160 | 0.100* |
| H18C | 0.1144 | 0.1031 | 0.5828 | 0.100* |
| C19 | 0.4068 (4) | 0.23638 (13) | 0.1148 (7) | 0.0655 (14) |
| H19A | 0.4793 | 0.2487 | 0.1237 | 0.098* |
| H19B | 0.3549 | 0.2570 | 0.1655 | 0.098* |
| H19C | 0.3893 | 0.2310 | -0.0109 | 0.098* |
| C20 | -0.0921 (4) | 0.10193 (18) | 0.5024 (7) | 0.0623 (13) |
| C21 | -0.0981 (4) | 0.0586 (2) | 0.5791 (8) | 0.0827 (16) |
| H21A | -0.0702 | 0.0326 | 0.5279 | 0.099* |
| C22 | -0.1507 (5) | 0.0618 (3) | 0.7375 (10) | 0.100 (2) |
| H22A | -0.1647 | 0.0382 | 0.8163 | 0.119* |
| C23 | -0.1448 (5) | 0.1283 (2) | 0.6193 (9) | 0.0912 (18) |
| H23A | -0.1557 | 0.1587 | 0.6025 | 0.109* |
| C24 | 0.4794 (4) | 0.09201 (16) | -0.2885 (7) | 0.0582 (12) |
| C25 | 0.5585 (4) | 0.05754 (19) | -0.3489 (8) | 0.0854 (17) |
| H25A | 0.5334 | 0.0444 | -0.4606 | 0.128* |
| H25B | 0.5645 | 0.0349 | -0.2570 | 0.128* |
| H25C | 0.6290 | 0.0710 | -0.3680 | 0.128* |
| C26 | 0.3122 (4) | 0.03785 (17) | 0.0706 (9) | 0.0687 (14) |
| C27 | 0.3269 (5) | 0.00620 (17) | 0.2258 (8) | 0.0891 (17) |
| H27A | 0.2609 | -0.0108 | 0.2424 | 0.134* |

| | | | | |
|------|------------|--------------|-------------|-------------|
| H27B | 0.3425 | 0.0227 | 0.3350 | 0.134* |
| H27C | 0.3868 | -0.0136 | 0.1999 | 0.134* |
| C28 | 0.6329 (4) | 0.18517 (16) | -0.0557 (7) | 0.0642 (14) |
| H28A | 0.7092 | 0.1930 | -0.0510 | 0.096* |
| H28B | 0.5900 | 0.2114 | -0.0785 | 0.096* |
| H28D | 0.6211 | 0.1640 | -0.1517 | 0.096* |
| C29 | 0.6656 (3) | 0.12137 (14) | 0.1587 (7) | 0.0639 (14) |
| H29A | 0.7420 | 0.1276 | 0.1397 | 0.096* |
| H29D | 0.6420 | 0.0989 | 0.0741 | 0.096* |
| H29B | 0.6548 | 0.1109 | 0.2810 | 0.096* |
| C30 | 0.2062 (4) | 0.18580 (15) | -0.0755 (7) | 0.0623 (13) |
| H30D | 0.2742 | 0.1970 | -0.1234 | 0.093* |
| H30A | 0.1640 | 0.2101 | -0.0270 | 0.093* |
| H30B | 0.1657 | 0.1715 | -0.1712 | 0.093* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| O1 | 0.060 (2) | 0.090 (2) | 0.103 (3) | -0.0292 (19) | -0.007 (2) | -0.020 (2) |
| O2 | 0.083 (2) | 0.112 (3) | 0.098 (3) | -0.040 (2) | 0.014 (2) | -0.026 (3) |
| O3 | 0.102 (3) | 0.148 (4) | 0.081 (3) | -0.035 (3) | 0.029 (3) | -0.017 (3) |
| O4 | 0.0501 (18) | 0.126 (3) | 0.069 (2) | -0.006 (2) | -0.0200 (18) | 0.018 (2) |
| O5 | 0.0452 (16) | 0.0519 (16) | 0.0514 (18) | -0.0011 (13) | 0.0003 (14) | -0.0054 (15) |
| O6 | 0.078 (2) | 0.089 (2) | 0.052 (2) | -0.001 (2) | -0.0079 (19) | -0.0021 (19) |
| O7 | 0.0501 (15) | 0.0375 (16) | 0.0581 (18) | -0.0025 (13) | -0.0017 (15) | 0.0014 (15) |
| O8 | 0.137 (4) | 0.066 (2) | 0.098 (3) | -0.015 (2) | -0.007 (3) | -0.027 (2) |
| C1 | 0.065 (3) | 0.052 (3) | 0.056 (3) | -0.011 (2) | 0.001 (2) | -0.011 (2) |
| C2 | 0.072 (3) | 0.063 (3) | 0.054 (3) | -0.021 (3) | -0.012 (3) | -0.004 (3) |
| C3 | 0.058 (3) | 0.051 (3) | 0.067 (3) | -0.014 (2) | -0.013 (3) | 0.000 (3) |
| C4 | 0.043 (2) | 0.046 (2) | 0.054 (3) | -0.007 (2) | 0.001 (2) | 0.002 (2) |
| C5 | 0.041 (2) | 0.039 (2) | 0.044 (2) | -0.0046 (18) | -0.003 (2) | 0.000 (2) |
| C6 | 0.046 (2) | 0.037 (2) | 0.045 (2) | 0.0048 (19) | 0.0000 (19) | -0.002 (2) |
| C7 | 0.040 (2) | 0.048 (2) | 0.048 (3) | -0.0015 (19) | -0.003 (2) | 0.004 (2) |
| C8 | 0.043 (2) | 0.047 (2) | 0.046 (2) | 0.004 (2) | -0.005 (2) | 0.003 (2) |
| C9 | 0.049 (2) | 0.045 (2) | 0.052 (3) | 0.001 (2) | 0.000 (2) | -0.004 (2) |
| C10 | 0.047 (2) | 0.038 (2) | 0.047 (3) | -0.0009 (19) | -0.005 (2) | -0.004 (2) |
| C11 | 0.065 (3) | 0.058 (3) | 0.075 (3) | 0.001 (2) | 0.009 (3) | -0.016 (3) |
| C12 | 0.059 (3) | 0.068 (3) | 0.072 (3) | 0.004 (2) | 0.006 (3) | -0.008 (3) |
| C13 | 0.040 (2) | 0.058 (3) | 0.055 (3) | 0.002 (2) | -0.005 (2) | -0.003 (2) |
| C14 | 0.043 (2) | 0.064 (3) | 0.051 (3) | 0.002 (2) | -0.006 (2) | 0.001 (2) |
| C15 | 0.059 (3) | 0.105 (5) | 0.065 (3) | -0.023 (3) | 0.001 (3) | -0.028 (3) |
| C16 | 0.049 (3) | 0.093 (4) | 0.069 (3) | -0.018 (3) | 0.005 (3) | -0.014 (3) |
| C17 | 0.037 (2) | 0.073 (3) | 0.062 (3) | 0.002 (2) | -0.001 (2) | 0.000 (3) |
| C18 | 0.050 (3) | 0.089 (4) | 0.062 (3) | -0.005 (3) | -0.006 (2) | 0.018 (3) |
| C19 | 0.066 (3) | 0.037 (2) | 0.093 (4) | -0.003 (2) | 0.001 (3) | 0.006 (3) |
| C20 | 0.042 (2) | 0.078 (4) | 0.067 (3) | -0.003 (3) | 0.000 (2) | -0.005 (3) |
| C21 | 0.066 (3) | 0.098 (4) | 0.084 (4) | -0.003 (3) | 0.003 (3) | 0.015 (4) |
| C22 | 0.078 (4) | 0.130 (6) | 0.091 (5) | -0.033 (4) | -0.004 (4) | 0.028 (5) |

supplementary materials

| | | | | | | |
|-----|-----------|-----------|-----------|------------|------------|------------|
| C23 | 0.088 (4) | 0.101 (4) | 0.085 (4) | -0.019 (3) | 0.037 (4) | -0.013 (4) |
| C24 | 0.052 (3) | 0.064 (3) | 0.058 (3) | -0.010 (2) | 0.002 (3) | -0.010 (3) |
| C25 | 0.086 (4) | 0.088 (4) | 0.082 (4) | 0.005 (3) | 0.010 (3) | -0.030 (3) |
| C26 | 0.065 (3) | 0.051 (3) | 0.090 (4) | -0.007 (2) | -0.007 (3) | -0.010 (3) |
| C27 | 0.092 (4) | 0.054 (3) | 0.122 (4) | -0.004 (3) | -0.009 (4) | 0.020 (4) |
| C28 | 0.055 (3) | 0.069 (3) | 0.069 (3) | -0.013 (2) | -0.001 (2) | 0.013 (3) |
| C29 | 0.048 (2) | 0.057 (3) | 0.086 (4) | 0.001 (2) | -0.013 (3) | -0.001 (3) |
| C30 | 0.058 (3) | 0.060 (3) | 0.069 (3) | 0.010 (2) | -0.003 (2) | 0.006 (3) |

Geometric parameters (Å, °)

| | | | |
|----------|-----------|----------|-----------|
| O1—C3 | 1.224 (5) | C12—H12B | 0.9700 |
| O2—C16 | 1.200 (6) | C13—C18 | 1.520 (6) |
| O3—C22 | 1.359 (8) | C13—C14 | 1.532 (6) |
| O3—C23 | 1.360 (7) | C13—C17 | 1.573 (6) |
| O4—C15 | 1.424 (7) | C14—C15 | 1.464 (6) |
| O4—C14 | 1.440 (5) | C15—C16 | 1.499 (7) |
| O5—C24 | 1.362 (5) | C15—H15A | 0.9800 |
| O5—C6 | 1.453 (5) | C16—C17 | 1.506 (7) |
| O6—C24 | 1.199 (5) | C17—C20 | 1.500 (7) |
| O7—C26 | 1.354 (6) | C17—H17A | 0.9800 |
| O7—C7 | 1.442 (5) | C18—H18A | 0.9600 |
| O8—C26 | 1.195 (6) | C18—H18B | 0.9600 |
| C1—C2 | 1.319 (6) | C18—H18C | 0.9600 |
| C1—C10 | 1.516 (6) | C19—H19A | 0.9600 |
| C1—H1A | 0.9300 | C19—H19B | 0.9600 |
| C2—C3 | 1.447 (7) | C19—H19C | 0.9600 |
| C2—H2A | 0.9300 | C20—C23 | 1.331 (7) |
| C3—C4 | 1.522 (6) | C20—C21 | 1.419 (7) |
| C4—C29 | 1.546 (6) | C21—C22 | 1.331 (8) |
| C4—C28 | 1.547 (6) | C21—H21A | 0.9300 |
| C4—C5 | 1.564 (5) | C22—H22A | 0.9300 |
| C5—C6 | 1.532 (5) | C23—H23A | 0.9300 |
| C5—C10 | 1.560 (5) | C24—C25 | 1.482 (7) |
| C5—H5A | 0.9800 | C25—H25A | 0.9600 |
| C6—C7 | 1.528 (5) | C25—H25B | 0.9600 |
| C6—H6A | 0.9800 | C25—H25C | 0.9600 |
| C7—C8 | 1.523 (6) | C26—C27 | 1.493 (7) |
| C7—H7A | 0.9800 | C27—H27A | 0.9600 |
| C8—C30 | 1.545 (6) | C27—H27B | 0.9600 |
| C8—C9 | 1.559 (6) | C27—H27C | 0.9600 |
| C8—C14 | 1.564 (6) | C28—H28A | 0.9600 |
| C9—C11 | 1.532 (6) | C28—H28B | 0.9600 |
| C9—C10 | 1.568 (6) | C28—H28D | 0.9600 |
| C9—H9A | 0.9800 | C29—H29A | 0.9600 |
| C10—C19 | 1.546 (6) | C29—H29D | 0.9600 |
| C11—C12 | 1.531 (6) | C29—H29B | 0.9600 |
| C11—H11A | 0.9700 | C30—H30D | 0.9600 |
| C11—H11B | 0.9700 | C30—H30A | 0.9600 |

| | | | |
|------------|-----------|---------------|-----------|
| C12—C13 | 1.529 (6) | C30—H30B | 0.9600 |
| C12—H12A | 0.9700 | | |
| C22—O3—C23 | 106.2 (5) | C15—C14—C8 | 128.1 (4) |
| C15—O4—C14 | 61.5 (3) | C13—C14—C8 | 120.8 (3) |
| C24—O5—C6 | 116.4 (3) | O4—C15—C14 | 59.8 (3) |
| C26—O7—C7 | 118.8 (4) | O4—C15—C16 | 107.3 (5) |
| C2—C1—C10 | 122.2 (4) | C14—C15—C16 | 107.0 (4) |
| C2—C1—H1A | 118.9 | O4—C15—H15A | 122.2 |
| C10—C1—H1A | 118.9 | C14—C15—H15A | 122.2 |
| C1—C2—C3 | 121.1 (5) | C16—C15—H15A | 122.2 |
| C1—C2—H2A | 119.4 | O2—C16—C15 | 126.2 (5) |
| C3—C2—H2A | 119.4 | O2—C16—C17 | 127.5 (5) |
| O1—C3—C2 | 120.8 (5) | C15—C16—C17 | 106.3 (4) |
| O1—C3—C4 | 120.2 (5) | C20—C17—C16 | 116.2 (4) |
| C2—C3—C4 | 118.9 (4) | C20—C17—C13 | 116.6 (4) |
| C3—C4—C29 | 103.2 (4) | C16—C17—C13 | 101.8 (3) |
| C3—C4—C28 | 110.0 (3) | C20—C17—H17A | 107.2 |
| C29—C4—C28 | 108.4 (4) | C16—C17—H17A | 107.2 |
| C3—C4—C5 | 111.1 (4) | C13—C17—H17A | 107.2 |
| C29—C4—C5 | 109.4 (3) | C13—C18—H18A | 109.5 |
| C28—C4—C5 | 114.1 (4) | C13—C18—H18B | 109.5 |
| C6—C5—C10 | 109.8 (3) | H18A—C18—H18B | 109.5 |
| C6—C5—C4 | 115.2 (3) | C13—C18—H18C | 109.5 |
| C10—C5—C4 | 114.2 (3) | H18A—C18—H18C | 109.5 |
| C6—C5—H5A | 105.6 | H18B—C18—H18C | 109.5 |
| C10—C5—H5A | 105.6 | C10—C19—H19A | 109.5 |
| C4—C5—H5A | 105.6 | C10—C19—H19B | 109.5 |
| O5—C6—C7 | 107.4 (3) | H19A—C19—H19B | 109.5 |
| O5—C6—C5 | 109.1 (3) | C10—C19—H19C | 109.5 |
| C7—C6—C5 | 112.1 (3) | H19A—C19—H19C | 109.5 |
| O5—C6—H6A | 109.4 | H19B—C19—H19C | 109.5 |
| C7—C6—H6A | 109.4 | C23—C20—C21 | 105.3 (5) |
| C5—C6—H6A | 109.4 | C23—C20—C17 | 126.1 (5) |
| O7—C7—C8 | 108.8 (3) | C21—C20—C17 | 128.7 (5) |
| O7—C7—C6 | 108.2 (3) | C22—C21—C20 | 107.7 (6) |
| C8—C7—C6 | 110.5 (3) | C22—C21—H21A | 126.1 |
| O7—C7—H7A | 109.8 | C20—C21—H21A | 126.1 |
| C8—C7—H7A | 109.8 | C21—C22—O3 | 109.6 (6) |
| C6—C7—H7A | 109.8 | C21—C22—H22A | 125.2 |
| C7—C8—C30 | 107.9 (3) | O3—C22—H22A | 125.2 |
| C7—C8—C9 | 109.7 (3) | C20—C23—O3 | 111.1 (6) |
| C30—C8—C9 | 113.6 (3) | C20—C23—H23A | 124.4 |
| C7—C8—C14 | 111.9 (3) | O3—C23—H23A | 124.4 |
| C30—C8—C14 | 107.3 (3) | O6—C24—O5 | 122.8 (4) |
| C9—C8—C14 | 106.5 (3) | O6—C24—C25 | 125.9 (5) |
| C11—C9—C8 | 110.8 (3) | O5—C24—C25 | 111.3 (5) |
| C11—C9—C10 | 115.8 (3) | C24—C25—H25A | 109.5 |
| C8—C9—C10 | 114.7 (3) | C24—C25—H25B | 109.5 |
| C11—C9—H9A | 104.7 | H25A—C25—H25B | 109.5 |

supplementary materials

| | | | |
|---------------|------------|-----------------|------------|
| C8—C9—H9A | 104.7 | C24—C25—H25C | 109.5 |
| C10—C9—H9A | 104.7 | H25A—C25—H25C | 109.5 |
| C1—C10—C19 | 105.8 (3) | H25B—C25—H25C | 109.5 |
| C1—C10—C5 | 104.6 (3) | O8—C26—O7 | 122.5 (5) |
| C19—C10—C5 | 115.4 (4) | O8—C26—C27 | 127.1 (5) |
| C1—C10—C9 | 108.5 (3) | O7—C26—C27 | 110.4 (5) |
| C19—C10—C9 | 114.7 (3) | C26—C27—H27A | 109.5 |
| C5—C10—C9 | 107.2 (3) | C26—C27—H27B | 109.5 |
| C12—C11—C9 | 114.3 (4) | H27A—C27—H27B | 109.5 |
| C12—C11—H11A | 108.7 | C26—C27—H27C | 109.5 |
| C9—C11—H11A | 108.7 | H27A—C27—H27C | 109.5 |
| C12—C11—H11B | 108.7 | H27B—C27—H27C | 109.5 |
| C9—C11—H11B | 108.7 | C4—C28—H28A | 109.5 |
| H11A—C11—H11B | 107.6 | C4—C28—H28B | 109.5 |
| C13—C12—C11 | 114.6 (4) | H28A—C28—H28B | 109.5 |
| C13—C12—H12A | 108.6 | C4—C28—H28D | 109.5 |
| C11—C12—H12A | 108.6 | H28A—C28—H28D | 109.5 |
| C13—C12—H12B | 108.6 | H28B—C28—H28D | 109.5 |
| C11—C12—H12B | 108.6 | C4—C29—H29A | 109.5 |
| H12A—C12—H12B | 107.6 | C4—C29—H29D | 109.5 |
| C18—C13—C12 | 113.7 (4) | H29A—C29—H29D | 109.5 |
| C18—C13—C14 | 113.5 (4) | C4—C29—H29B | 109.5 |
| C12—C13—C14 | 106.4 (4) | H29A—C29—H29B | 109.5 |
| C18—C13—C17 | 106.4 (4) | H29D—C29—H29B | 109.5 |
| C12—C13—C17 | 114.6 (4) | C8—C30—H30D | 109.5 |
| C14—C13—C17 | 101.8 (3) | C8—C30—H30A | 109.5 |
| O4—C14—C15 | 58.7 (3) | H30D—C30—H30A | 109.5 |
| O4—C14—C13 | 110.9 (3) | C8—C30—H30B | 109.5 |
| C15—C14—C13 | 108.5 (4) | H30D—C30—H30B | 109.5 |
| O4—C14—C8 | 112.2 (3) | H30A—C30—H30B | 109.5 |
| C10—C1—C2—C3 | -0.6 (7) | C11—C12—C13—C14 | -47.8 (5) |
| C1—C2—C3—O1 | 155.2 (5) | C11—C12—C13—C17 | -159.4 (4) |
| C1—C2—C3—C4 | -27.8 (7) | C15—O4—C14—C13 | -99.5 (4) |
| O1—C3—C4—C29 | 68.6 (5) | C15—O4—C14—C8 | 122.1 (4) |
| C2—C3—C4—C29 | -108.4 (4) | C18—C13—C14—O4 | 151.9 (4) |
| O1—C3—C4—C28 | -46.8 (6) | C12—C13—C14—O4 | -82.3 (4) |
| C2—C3—C4—C28 | 136.1 (4) | C17—C13—C14—O4 | 37.9 (5) |
| O1—C3—C4—C5 | -174.2 (4) | C18—C13—C14—C15 | 89.2 (4) |
| C2—C3—C4—C5 | 8.8 (6) | C12—C13—C14—C15 | -145.0 (4) |
| C3—C4—C5—C6 | 162.7 (3) | C17—C13—C14—C15 | -24.8 (5) |
| C29—C4—C5—C6 | -84.1 (4) | C18—C13—C14—C8 | -73.9 (5) |
| C28—C4—C5—C6 | 37.6 (5) | C12—C13—C14—C8 | 51.9 (5) |
| C3—C4—C5—C10 | 34.3 (5) | C17—C13—C14—C8 | 172.1 (4) |
| C29—C4—C5—C10 | 147.6 (4) | C7—C8—C14—O4 | -108.7 (4) |
| C28—C4—C5—C10 | -90.8 (4) | C30—C8—C14—O4 | 9.4 (5) |
| C24—O5—C6—C7 | 82.0 (4) | C9—C8—C14—O4 | 131.4 (4) |
| C24—O5—C6—C5 | -156.3 (3) | C7—C8—C14—C15 | -41.9 (6) |
| C10—C5—C6—O5 | 179.4 (3) | C30—C8—C14—C15 | 76.3 (6) |
| C4—C5—C6—O5 | 48.8 (4) | C9—C8—C14—C15 | -161.8 (5) |

| | | | |
|-----------------|------------|-----------------|------------|
| C10—C5—C6—C7 | -61.8 (4) | C7—C8—C14—C13 | 117.6 (4) |
| C4—C5—C6—C7 | 167.7 (3) | C30—C8—C14—C13 | -124.2 (4) |
| C26—O7—C7—C8 | 136.6 (4) | C9—C8—C14—C13 | -2.3 (5) |
| C26—O7—C7—C6 | -103.4 (4) | C14—O4—C15—C16 | 100.0 (4) |
| O5—C6—C7—O7 | 60.8 (4) | C13—C14—C15—O4 | 103.7 (4) |
| C5—C6—C7—O7 | -59.0 (4) | C8—C14—C15—O4 | -94.8 (5) |
| O5—C6—C7—C8 | 179.8 (3) | O4—C14—C15—C16 | -100.5 (5) |
| C5—C6—C7—C8 | 60.1 (4) | C13—C14—C15—C16 | 3.2 (6) |
| O7—C7—C8—C30 | -171.2 (3) | C8—C14—C15—C16 | 164.7 (4) |
| C6—C7—C8—C30 | 70.2 (4) | O4—C15—C16—O2 | 137.6 (6) |
| O7—C7—C8—C9 | 64.6 (4) | C14—C15—C16—O2 | -159.5 (5) |
| C6—C7—C8—C9 | -54.0 (4) | O4—C15—C16—C17 | -41.7 (5) |
| O7—C7—C8—C14 | -53.4 (4) | C14—C15—C16—C17 | 21.2 (6) |
| C6—C7—C8—C14 | -172.0 (3) | O2—C16—C17—C20 | 17.0 (8) |
| C7—C8—C9—C11 | -172.5 (3) | C15—C16—C17—C20 | -163.7 (4) |
| C30—C8—C9—C11 | 66.8 (5) | O2—C16—C17—C13 | 144.8 (5) |
| C14—C8—C9—C11 | -51.2 (4) | C15—C16—C17—C13 | -35.9 (5) |
| C7—C8—C9—C10 | 54.2 (4) | C18—C13—C17—C20 | 44.8 (5) |
| C30—C8—C9—C10 | -66.6 (5) | C12—C13—C17—C20 | -81.8 (5) |
| C14—C8—C9—C10 | 175.5 (3) | C14—C13—C17—C20 | 163.9 (4) |
| C2—C1—C10—C19 | -80.6 (5) | C18—C13—C17—C16 | -82.7 (5) |
| C2—C1—C10—C5 | 41.7 (5) | C12—C13—C17—C16 | 150.7 (4) |
| C2—C1—C10—C9 | 155.8 (4) | C14—C13—C17—C16 | 36.4 (5) |
| C6—C5—C10—C1 | 171.7 (3) | C16—C17—C20—C23 | -153.2 (5) |
| C4—C5—C10—C1 | -57.3 (4) | C13—C17—C20—C23 | 86.7 (6) |
| C6—C5—C10—C19 | -72.5 (4) | C16—C17—C20—C21 | 25.9 (7) |
| C4—C5—C10—C19 | 58.5 (5) | C13—C17—C20—C21 | -94.2 (6) |
| C6—C5—C10—C9 | 56.6 (4) | C23—C20—C21—C22 | -1.6 (6) |
| C4—C5—C10—C9 | -172.3 (3) | C17—C20—C21—C22 | 179.0 (5) |
| C11—C9—C10—C1 | 61.7 (5) | C20—C21—C22—O3 | 0.9 (6) |
| C8—C9—C10—C1 | -167.3 (3) | C23—O3—C22—C21 | 0.2 (7) |
| C11—C9—C10—C19 | -56.3 (5) | C21—C20—C23—O3 | 1.8 (6) |
| C8—C9—C10—C19 | 74.7 (4) | C17—C20—C23—O3 | -178.9 (4) |
| C11—C9—C10—C5 | 174.2 (4) | C22—O3—C23—C20 | -1.3 (7) |
| C8—C9—C10—C5 | -54.8 (4) | C6—O5—C24—O6 | -3.1 (6) |
| C8—C9—C11—C12 | 55.8 (5) | C6—O5—C24—C25 | 175.7 (3) |
| C10—C9—C11—C12 | -171.4 (4) | C7—O7—C26—O8 | -0.7 (7) |
| C9—C11—C12—C13 | -2.1 (6) | C7—O7—C26—C27 | 178.6 (4) |
| C11—C12—C13—C18 | 77.9 (5) | | |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|-----------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C5—H5A \cdots O7 | 0.98 | 2.45 | 2.900 (5) | 107 |
| C7—H7A \cdots O8 | 0.98 | 2.31 | 2.705 (5) | 103 |
| C9—H9A \cdots O7 | 0.98 | 2.54 | 2.936 (5) | 104 |
| C27—H27A \cdots O2 ⁱ | 0.96 | 2.51 | 3.422 (7) | 160 |
| C28—H28D \cdots O5 | 0.96 | 2.53 | 3.121 (6) | 120 |
| C29—H29D \cdots O5 | 0.96 | 2.29 | 2.973 (5) | 127 |

supplementary materials

Symmetry codes: (i) $-x, -y, z$.

Fig. 1

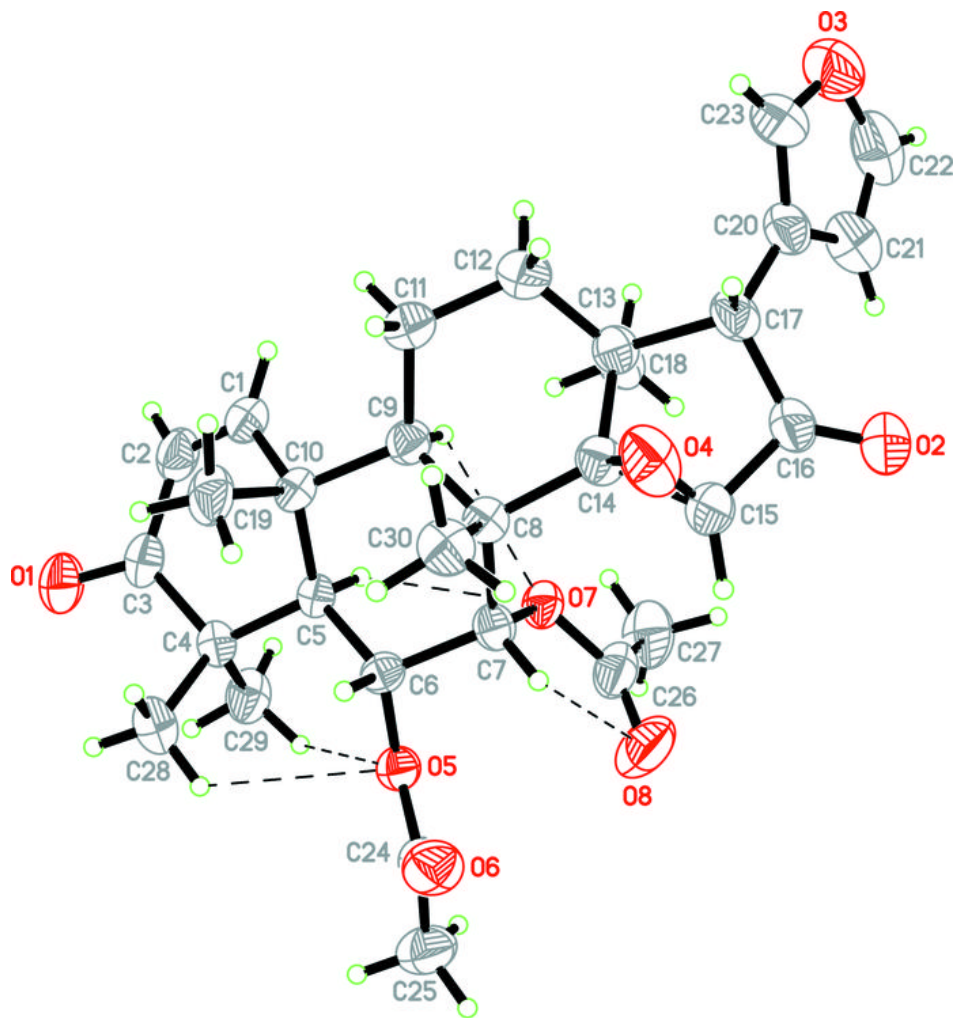


Fig. 2

